Science Series Part 2:

Get in "The Zone" – A Simple Approach to Soil Test Analysis

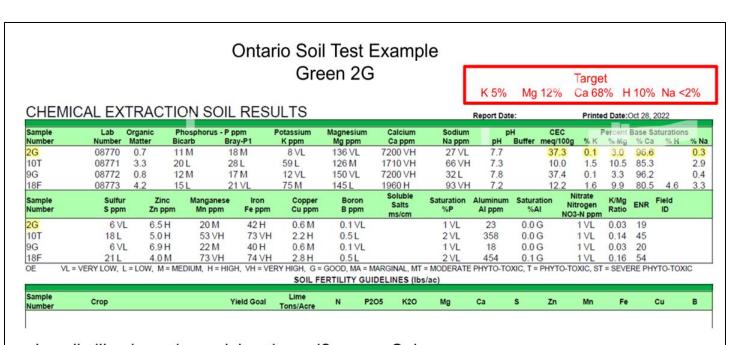
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Soil Test & Saturated Paste Analysis Test Case Study Examples

Understanding each element's role in the turfgrass system is crucial for interpreting a soil analysis. By knowing how these elements are quantified in various tests and focusing on a few key values, you should be able to interpret a soil test and paste extract in just a few minutes to determine what actions, if any, can improve nutrient efficiency, thus enhancing performance conditions at your golf course.

The following three test result examples are paired with recommended solutions to address the identified issues

EXAMPLE #1



Is soil silica based or calcium based?

Is Sodium OK?

Is Calcium OK?

Is Potassium OK?

Is Magnesium OK?

Calcareous

Sodium < 2% OK

Calcium > 68% high (calcareous)

Potassium < 5%, low BUT

Magnesium < 12%, low BUT

Need more potassium in the soil

Reported Date(2022-11-09 Printed Date(Nov 9, 2022

Saturated Paste Analysis

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Ontario Soil Test Example

PASTE EXTRACTION SOIL RESULTS

Green 2G

Sample Number	Lab Number	Organic Metter	pH	pH Duffer	Phosphorus P ppm	Potassium K ppm	Magnesium Mg ppm	Calcium Ca ppm	Sulfur 5 ppm	Sodium Na ppm	Soluble Salts ms/om	Nitrate Nitrogen NO3-N ppm	Chlorida Clippen
23	12489		7.8		2.244	4	4	12	2	13	0.14	3	7
10T	12490		7.5		2.91	9	6	38	11	22	0.3	3	36
9G	12491		7.9		1.003	4	3.	13	2	11	0.12	3	7
1.0F	12492		7.5		2.411	9	7	41	19	30	0.33	3	41
Optimum R	tange				7-12	30-40	2/0-2/5	125-200	12-15	4-10			
Sample	Lab	Zime	Managanese	le per	Copper	Doron	PROD-N		Percent Dase	Salarations			
Mumber	Mumber	Zn.ppm	Mn ppm	Fe ppm	CHARDIN	в ррен	200	76.8	75 Mg	75 CA	N. Ma	3/CI	3/403
20	12409		1	1	0.029	0.022	3	5.2	22.7	34	30.4	7.1	3.1
10T	12490		1	2	0.047	0.011	3	6	13.9	49.9	30.4	17.1	1.4
90	12491	0.026	1	1	0.017	0.023	3	6.1	19.9	39.7	34.5	0.3	3.6
18F	12492	0.02	1	2	0.044	0.015	3	5.9	14.5	43.4	36.4	17.7	1.3
Optimum R	lange							11-13	8-15	55-75	410	410	8-19

2G

1. Wilt Susceptibility

- K% > Na%
- 5.2 < 38.4

Turf susceptible to wilt

2. Ca% : Mg% + K% Ratio

Ca > Mg + K

34 Ca > 22.7 Mg + 5.2 K

Ca: Mg& K Ratio GOOD

Need potassium

Increase solubility with acids

3. Enough Total Soluble Ca?

Ca > 40 ppm

Ca = 12 ppm

Need more soluble Ca. Solubilize with acids

Summary for Green 2G

This calcareous soil is adequate, but it tends to wilt during the growing season due to the high sodium-to-potassium ratio in the saturated paste test. The low available calcium (12 ppm) results in a softer playing surface because of the elevated soluble sodium.

<u>Action</u>

- 1. Increasing soluble calcium by applying a long-lasting acid will help displace sodium and firm the surface.
- 2. Adding potassium before heat or traffic stress will minimize sodium-induced wilt.

EXAMPLE #2

Job N	lame: Alberta Exa	mple	Soil Rep	oort	9/26/202	2	
Sample	e Location		Green	Green	Green		
Sample	e ID		7	10	13		
Lab No	umber		36	37	38		
Sample	e Depth in Inches		8	8	8		
Total E	Exchange Capacity (M. E.)		11.75	15.29	14.63		
pH of S	Soll Sample		8.7	8.6	8.6		
Organi	c Matter, Percent		0.29	0.58	0.65		
SNO	SULFUR:	p.p.m.	13	12	14		
ANIONS	Mehlich III Phosphorous:		17	17	22		
EXCHANGE ABLE CATIONS	CALCIUM: ppm	Desired Value Value Found Deficit	1597 2004	2079 2628	1989 2528		
	MAGNESIUM: ppm	Desired Value Value Found Deficit	169 143 -26	220 173 -47	210 156 -54		
EXCHAN	POTASSIUM: ppm	Desired Value Value Found Deficit	183 44 -139	238 68 -170	228 71 -157		
	SODIUM:	ppm	24	24	21		
3º	Calcium (60 to 70%)	85.30	85.94	86.43			
NO.	Magnesium (10 to 20%)	10.16	9.42	8.89			
R	Potassium (2 to 5%)	0.96	1.14	1.25			
ATU	Sodium (.5 to 3%)	0.89	0.69	0.64			
BASE SATURATION	Other Bases (Variable)	2.70	2.80	2.80			
9	Exchangable Hydrogen (10 to 1	0.00	0.00	0.00			
S	Boron (p.p.m.)	0.71	0.46	0.54			
RACE ELEMENTS	Iron (p.p.m.)	110	127	130			
	Manganese (p.p.m.)	53	54	53			
	Copper (p.p.m.)	< 0.2	0.25	0.23			
	Zinc (p.p.m.)	0.63	0.87	0.78			
_	Aluminum (p.p.m.)	38	45	46			
OTHER	Ammonium (p.p.m.) Nitrate (p.p.m.)	0.5	0.4 1.5	0.5			
D				-			

Green 13

Is soil silica based or calcium based? Is Sodium OK?

Is Calcium OK?

Is Potassium OK?

Is Magnesium OK?

Calcareous (CEC 14.63) pH 8.6 Sodium < 2% OK (actual 0.64) Calcium > 68% high (calcareous) OK actual 86.43 Potassium < 5%, low BUT Should be at least 1.28-Need potassium Magnesium < 8.89%, Add a little

Note ammonium ppm vs nitrate ppm Take soil profile

		Satur	ated Pas	te Repoi	t		
Jòó Nam	Alberta Example				9/26/202	2	
,,							
Sample	Location		Green	Green	Green		
Sample	ID		7	10	13		
Lub Nun	mber		187086	187087	187088		
Water U	Ised		DI	DI	DI		
рН			8.7	8.6	8.6		
Soluble	Salts	ppm	60	52	58		
Chloride	+ (a)	ppm	11	8	7		
Bicarbon	nute (HCO3)	ppm	44	34	31		
	SULFUR	ppm	2.19	2	2.65		
ANIONS	PHOSPHORUS	ppm	0.83	0.68	0.7		
	CALCIUM	ppm	7.54	6.85	7.72		
		meq/l	0.38	0.34	0.39		
ONS	MAGNESIUM	ppm	3.53	2.88	3.17		
SOLUBLE CATIONS		meq/l	0.29	0.24	0.26		
層	POTASSIUM:	ppm	3.81	4.12	4.16		
SO	POTASSION	meq/l	0.10	0.11	0.11		
	SODIUM	ppm	3.88	3.07	3.32		
	3001011	meq/I	0.17	0.13	0.14		
	Calcium		40.18	41.63	42.80		
ENT	Magnesium		31.31	29.14	29.24		
PERCENT	Potassium	10.55	13.01	11.98			
-	Sodium		17.96	16.22	15.98		
S	Boron (p.p.m.)	0.05	0.04	0.03			
TRACE ELEMENTS	Iron (p.p.m.)	3.09	2.51	3.12			
	Manganese (p.p.m.)	0.08	0.05	0.06			
	Copper (p.p.m.)	< 0.02	< 0.02	< 0.02			
	Zinc (p.p.m.)	< 0.02	< 0.02	< 0.02			
	Aluminum (p.p.m.)	0.52	0.57	0.74			
or.							
ОТНЕВ							
0							

GREEN 13

- 1. Wilt Susceptibility
 - K% > Na%
 - 11.98 < 15.98

Turf susceptible to wilt

2. Ca% : Mg% + K% Ratio
Ca > Mg + K
42.8% Ca > 29.24 Mg + 11.98 K
Ca: Mg& K Ratio OK

3. Enough Total Soluble Ca? Ca > 40 ppm

Ca = 7.72 ppm

Need more soluble Ca.

Solubilize with acids

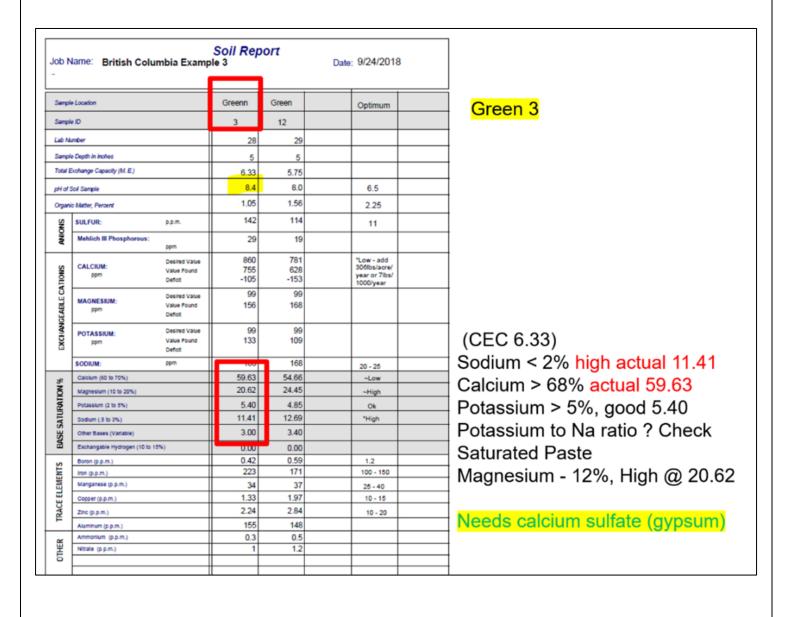
Summary for Green 13

This calcareous soil is adequate but slightly prone to wilting during the growing season due to the high sodium-to-potassium ratio from the saturated paste test. The low available calcium (7.72 ppm) will create a softer playing surface because of the increased soluble sodium.

Action

- 1. Increasing the soluble calcium with long-lasting acid will help displace and sink the sodium, thereby firming the surface.
- 2. Adding potassium before heat or traffic stress will minimize sodium-induced wilt.

EXAMPLE #3



Job Name	British Columbia I		ated Pas	te Repoi	t	9/24/2018	1	- GREEN 3
Sample	Sample Location Green Green					Optimum		
Simple	Semple ID 3 12						1. Wilt Susceptibility	
Lab Nun	Lab Number 123451 123452						• K% > Na%	
Water I	Water Used DI DI						• 7.28 < 42.11	
	sec.		8.4	8.0		6.5	*High	1
ρΗ							riigii	Prone to wilt. Decrease sodium
Soluble.	Salts	ppm	367	285		< 960		
Chloride	(a)	ppm	44	33		< 50		⊉. Ca% : Mg% + K% Ratio
Bicarbor	Sicarbonute (HC03) ppm			127		< 50	* High	Ca> Mg+K
SE .	SULFUR	ppm	40	27.05		5 - 10	*High	
ANIONS	PHOSPHORUS	ppm	0.84	0.69		1-3	~Low	32.54 <mark>% Ca > 18.07 Mg + 7.28 K</mark>
	CALCIUM	ppm	37.31	26.94		40 - 60	* Low	Ca: Mg & K ratio good
		meq/I	1.87	1.35				
SNO	MAGNESIUM	ppm	12.43	9.70		8 - 12		
E E		meq/I	1.04	0.81				
SOLUBLE CATIONS	POTASSIUM:	ppm	16.06	11.20		15 - 20		3. Enough Total Soluble Ca?
SOL		meq/I	0.42	0.29				
	SODIUM	ppm	55.53	46.03		< 20	* High	Ca > 40 ppm
		meq/I	2.41	2.00				Ca = 37.31 ppm
	Calcium	Calcium		30.29		55 - 60	*Low	Need more soluble Ca.
ENT	Magnesium		18.07	18.17		18 - 20	Ok	
PERCENT	Potassium		7.28	6.54		9 - 10	~Low	Solubilize with acids
	Sodium		42.11	45.00		< 8	*High	
IIS	Boron (p.p.m.)		0.02	0.02		0.1		
ELEMENTS	Iron (p.p.m.)		0.09	1.16 0.09		0.3		-
ELE	Manganese (p.p.m.) Copper (p.p.m.)		0.09	0.09		0.1		1
TRACE	Zinc (p.p.m.)		0.02	0.02		0.08		1
IR	Aluminum (p.p.m.)		2.13	1.34				1
ОТНЕЯ		2) (==) ((0 (00)					

Summary for Green 3

Sodium is very high at 11.41%. What is causing this? According to the chemical extraction test, calcium is low at 59.63%. However, another factor contributes to the elevated sodium levels.

The saturated paste test shows that soluble calcium is slightly low at 37.31 ppm, while the soluble sodium ppm exceeds that of calcium. Additionally, the bicarbonates are at 112 ppm, which is high since they should be around 50 ppm in the saturated paste test.

<u>Action</u>

- 1. Perform an irrigation water test to determine the source of the sodium and whether it contributes to the high bicarbonate level.
- 2. Regardless, the strategy to reduce sodium is to utilize calcium sulfate. The calcium will displace the sodium, while the sulfate will combine with the sodium, creating sodium sulfate, allowing the sodium to sink through the soil. Spray long-lasting acid surfactants to solubilize more calcium and displace the sodium. 3. Adding potassium before heat or traffic stress will minimize sodium-induced wilt.